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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.005 Å R factor = 0.058 wR factor = 0.139 Data-to-parameter ratio = 17.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

4-(2,3-Dichlorophenyl)-1-[3-oxo-3-(2-thienyl)propyl]piperazin-1-ium chloride

The title compound, $C_{17}H_{19}Cl_2N_2OS^+\cdot Cl^-$, was synthesized from 2-acetothiophene, 1-(2,3-dichlorophenyl)piperazine and paraformaldehyde. In the cation, the thienyl ring is coplanar with the oxopropyl group. The protonated piperazine ring exhibits a chair conformation. In the crystal packing the cations are connected by $C-H \cdot \cdot \cdot O$ and $C-H \cdot \cdot \cdot Cl$ hydrogen bonds.

Comment

3-(4-Arylpiperazin-1-yl)-1-arylpropane derivatives are a class of selective antidepressants with a low incidence of side effects. However, their therapeutic effect is achieved only after repeated administration (Oficialdegui *et al.*, 2000; Martinez *et al.*, 2001; Esparza *et al.*, 2001; Orus *et al.*, 2002).



The title compound, (I), was synthesized from 2acetothiophene, 1-(2,3-dichlorophenyl)piperazine and paraformaldehyde. The molecular structure of (I) is illustrated in Fig. 1. The thiopene ring is coplanar with the oxopropyl group (O1/C5/C6/C7), the average deviation of contributing atoms from the least-squares plane being 0.0172 (2) Å. The protonated piperazine ring is in a normal chair conformation. In the crystal packing, the cations are linked by C-H···O and C-H···Cl hydrogen bonds (Fig. 2 and Table 1).

Experimental

A mixture of 2-acetothiophene (15 mmol), 1-(2,3-dichlorophenyl)piperazine (15 mmol) and concentrated hydrochloric acid in absolute



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A view of (I), showing 30% probability displacement ellipsoids.

Received 11 October 2005 Accepted 11 November 2005 Online 16 November 2005 ethanol (20 ml) was refluxed at 351 K. Paraformaldehyde (45 mmol) was added over a period of 20 min and the mixture was refluxed for 12 h. The solution was cooled to room temperature, and acetone (20 ml) was added with stirring. The resulting solid was filtered off, washed with diethyl ether and dried in a vacuum to give the title compound. Crystals suitable for X-ray analysis were grown by slow evaporation of an absolute methanol solution at room temperature over a period of 15 days.

Mo $K\alpha$ radiation

reflections

 $\theta = 2.4-26.4^{\circ}$ $\mu = 0.59 \text{ mm}^{-1}$

T = 294 (2) K

 $R_{\rm int} = 0.037$ $\theta_{\rm max} = 26.4^{\circ}$

 $h = -11 \rightarrow 19$

 $k = -9 \rightarrow 8$

 $l = -42 \rightarrow 33$

Block, colourless

 $0.22\,\times\,0.18\,\times\,0.16$ mm

3934 independent reflections

2963 reflections with $I > 2\sigma(I)$

Cell parameters from 6615

Crystal data

 $\begin{array}{l} C_{17}H_{19}Cl_2N_2OS^+\cdot Cl^-\\ M_r = 405.75\\ Orthorhombic, Pbca\\ a = 15.400 \ (2) \ \text{\AA}\\ b = 7.4361 \ (11) \ \text{\AA}\\ c = 33.601 \ (5) \ \text{\AA}\\ V = 3847.9 \ (9) \ \text{\AA}^3\\ Z = 8\\ D_x = 1.401 \ \text{Mg m}^{-3} \end{array}$

Data collection

Bruker SMART 1000 CCD areadetector diffractometer φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 1997) $T_{\min} = 0.850, T_{\max} = 0.910$ 20192 measured reflections

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.044P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.058$	+ 4.8252P]
$wR(F^2) = 0.139$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.12	$(\Delta/\sigma)_{\rm max} = 0.001$
3934 reflections	$\Delta \rho_{\rm max} = 0.35 \ {\rm e} \ {\rm \AA}^{-3}$
221 parameters	$\Delta \rho_{\rm min} = -0.31 \text{ e } \text{\AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C11-H11B\cdots O1^{i}$	0.97	2.50	3.002 (4)	112
$C9-H9B\cdots Cl2^{ii}$	0.97	2.75	3.606 (3)	147

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) $x - \frac{1}{2}$, y, $-z + \frac{1}{2}$.

The H atom attached to the N atom was located in a difference density map and refined isotropically. All other H atoms were posi-



Figure 2

The crystal structure of (I), viewed along the *b* axis, showing the C– $H \cdot \cdot \cdot O$ and C– $H \cdot \cdot \cdot Cl$ hydrogen bonds (dashed lines). Cl[–] anions have been omitted for clarity.

tioned geometrically and refined as riding, with C-H = 0.93-0.97 Å and $U_{iso}(H) = 1.2 U_{co}(C)$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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